ASYMMETRIC SYNTHESIS OF R- β -AMINO BUTANOIC ACID AND S- β -TYROSINE: HOMOCHIRAL LITHIUM AMIDE EQUIVALENTS FOR MICHAEL ADDITIONS TO α , β -UNSATURATED ESTERS.

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(Received 14 February 1991)

Abstract: Michael addition of the lithium amide derived from R-N-(α -methylbenzyl)benzylamine to benzyl E-crotonate is highly stereoselective (95% d.e.) giving after debenzylation and crystallisation homochiral R- β -amino butanoic acid. A similar addition to methyl E-(pbenzyloxy)cinnamate is completely stereoselective giving after debenzylation and acid hydrolysis homochiral S- β -tyrosine as its HCl salt.

 β -Amino acids are constituents of many classes of natural products, for example, terpenes, alkaloids, peptides, β -lactam antibiotics, etc., as well as being important for the synthesis of compounds of pharmaceutical interest. Some homochiral (enantiomerically pure) β -amino acids have been made available *via* manipulation of the chiral pool,¹ while approaches to their asymmetric synthesis have, to date, generally been centred on chiral auxiliaries providing homochiral enolate and homochiral α , β -unsaturated acid equivalents for additions to imines² and Michael additions³ respectively.

Although the basic properties of homochiral lithium amides have been extensively studied for application to asymmetric synthesis their potential as homochiral nucleophiles in Michael additions has received little attention.⁴ The highly stereoselective (97% d.e.) Michael addition of the 1,1'-binaphthyl derived lithium amide of 3,5-dihydro-4*H*-dinaphth[2,1-c:1',2'-e]azepine to methyl crotonate has been described⁵, however, the subsequent release of the β -amino acid has not been reported.

We recently described good chiral recognition in the Michael addition reaction between lithium N-(3,4dimethoxybenzyl) α -methylbenzylamide and the chiral iron crotonyl complex E-[(C₅H₅)Fe(CO)(PPh₃)(COCH=CHMe)], which allowed the kinetic resolution of the latter.⁶ Implicit in the chiral recognition is that both the iron and the α -methylbenzyl chiral auxiliaries must each be exerting powerful stereochemical control over the formation of the β -chiral centre. We reasoned, therefore, that high stereoselectivities should be expected in the addition of lithium N-(3,4-dimethoxybenzyl) α -methylbenzylamide to simple E-crotonate esters. The Michael addition of α -methylbenzylamine to methyl crotonate in alcohol solvents proceeds in reasonable yield but with poor diastereoselectivity (6-20% d.e.).^{7,8} In our hands addition of R- α -methylbenzylamine 1 to methyl crotonate in ethanol at reflux gave the R,S- β -amino derivative 2 in 35% yield with <4% d.e. Addition of the lithium amide derived from 1 to methyl crotonate at -78°C in THF gave R,R-2, but with no discernible stereoselectivity (yield 28%; 0% d.e.): The low yield was due to extensive formation of N- α -methylbenzyl crotonamide (35%).



In contrast to the primary amine 1, the derived secondary amines 3, 4 and 5 do not yield any Michael adducts with methyl crotonate in ethanol at reflux. The corresponding lithium amides 6, 7 and 8 respectively, however, do add giving, after protonation, the Michael adducts in high yield and with high diastereoselectivity: All these additions were essentially instantaneous.



The diastereoselectivities and yields for the Michael additions of the lithium amides derived from R-3, R-4 and R,R-5 to alkyl crotonates are listed in the Table. In two cases the configuration of the major product diastereoisomer was unambiguously assigned as R,R, *vide infra*, and all others were assigned by analogy. Characteristically, in the ¹H n.m.r. spectra the β -methyl doublet for the major R,R-diastereoisomer in each case appeared between $\delta 1.0$ -1.2, whereas that of the minor R,S-diastereoisomer appeared between $\delta 0.8$ -0.9.

General procedure for the Michael additions: Butyl lithium (1.4M in hexanes; 2.0 mmol) was added to the secondary amine (2.0 mmol) in tetrahydrofuran (15ml) at 0°C. The resulting red solution was stirred at 0°C for 15 min and cooled to -78° C. The crotonate ester (1.0 mmol) in tetrahydrofuran (4ml) was added dropwise and the mixture stirred for 15 min at -78° C before quenching with saturated ammonium chloride (2 ml). The mixture was allowed to warm and poured into saturated aqueous sodium chloride solution. Extraction of the aqueous layer with ether (2 x 20 ml), drying (MgSO4), filtration and evaporation gave a mixture of the adduct and the excess amine as a pale yellow oil. ¹H n.m.r. spectroscopic analyses of these oils were used to determine the diastereomeric excesses. Column chromatography (silica: 40/60 petrol : ether, 4 : 1) gave the pure adducts as colourless oils with no change in d.e. Elemental microanalyses were obtained for the adducts or for the corresponding crystalline HCl salts.

Table

Diastereomeric excesses (and yields) for the addition of secondary lithium amides to E-crotonate esters



*>99% d.e. indicates that none of the minor diastereoisomer was detected **0°C (no reaction at -78°C)

The reaction of methyl crotonate with lithium amide 8 could be performed at 0°C without compromising the complete stereoselectivity while at 40°C the diastereomeric excess of the adduct dropped to 93%. Debenzylation of the above adducts could be achieved with Pearlman's catalyst under 1-5 atmospheres of hydrogen. Thus the hydrogenolysis of R,R-9 (95% d.e.; 516 mg, 1.33 mmol) in EtOH (15 ml) with Pd(OH)₂ on charcoal (20%) under *ca*. 5 atmospheres of hydrogen at 20°C for 22h gave, after removal of the catalyst by filtration and evaporation to dryness, β -amino butanoic acid as a colourless solid in quantitative yield. A single crystallisation from methanol gave homochiral R- β -amino butanoic acid R-10 (64%) [α]_D¹⁹ -39.8 (c = 0.47, H₂O) Lit.⁹ for S-10 [α]_D¹⁸ +38.8 (c = 0.48, H₂O).



Treatment of methyl *p*-benzyloxycinnamate 11 with the lithium amide 6 generated the corresponding adduct 12 as a single diastereoisomer. Debenzylation of 12 following a modification of a literature procedure⁸ (1.3g, 2.8 mmol) in MeOH (20ml), water (2ml) and acetic acid (0.5ml) in the presence of palladium hydroxide on charcoal (20%) under hydrogen (1 atm.) at 20°C for 18h gave, after filtration and evaporation, the

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homochiral methyl ester of S- β -tyrosine 13 as an oil in quantitative yield. The enantiomeric purity (>99%) was confirmed by ¹H n.m.r. spectroscopy in the presence of S-2,2,2-trifluoro-1-(9-anthryl)ethanol in comparison with a racernic sample. Compound 13 was fully characterised as the HCl salt $[\alpha]_D^{20} + 10.55$ (c = 1.9, H₂O). Homochiral S- β -tyrosine.HCl 14 $[\alpha]_D^{25} + 3.55$ (c = 1.38, H₂O)¹⁰ was available by acid hydrolysis (aq. HCl, reflux 16h) of the methyl ester 13. The absolute configuration of 14 was assigned by comparison with the literature data¹⁰ and hence the absolute configuration of 13 and relative configurations within 12 were unambiguously confirmed.



We thank Idemitsu Petrochemical Co., Ltd. (Japan) for support (to OI).

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